GLY 4200	
Homework Exercise 7	7

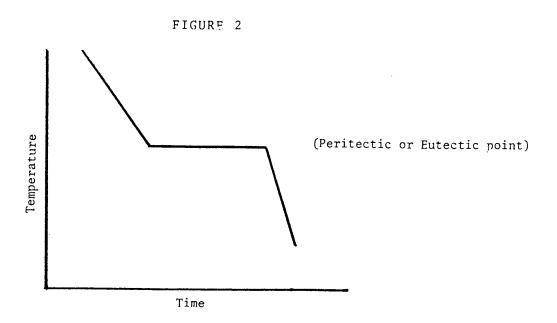
## **BINARY PHASE DIAGRAMS - PERITECTIC BEHAVIOR**

	shown?						
	b. How many phases are present in the region labeled $\beta$ -CaSiO <sub>3</sub> + -Walstromite						
	(BaCa2Si3O9)?						
	c. How many phases are present in the region labeled $\alpha$ -CaSiO <sub>3</sub> + Liq?						
	d. How many phases are there in the region labeled BaSiO <sub>3</sub> + Liq.?						
	e. How does the number of degrees of freedom change on going from $\alpha$ -CaSiO <sub>3</sub> + L						
	BaCa <sub>2</sub> Si <sub>3</sub> O <sub>9</sub> + Liq.?						
	f. How does the number of degrees of freedom change on going from to $BaCa_2Si_3Ca_2S$						
	Liq. to to $BaCa_2Si_3O_9 + BaSiO_{3?}$						
	g. List the phases present at the peritectic point.						
	h. How many degrees of freedom does the sample have at the peritectic point?						
•	i. List the phases present at the eutectic point.						
	j. How many degrees of freedom does the sample have at the eutectic point?						
	k. How do your answers to g through j explain the observed cooling curve (figure 2						
	which could apply to either the peritectic or eutectic points?						

- 2. Starting with a composition of 49% BaSiO<sub>3</sub>, trace the behavior of the melt from 1600°C to 1200°C. Show the path followed by the liquid in red, the path followed by the solid in blue.
  - a. At what temperature does the solid first appear?
  - b. What is the composition of the first solid?
  - c. At 1325°C, what phases are present?
  - d. What percent of each phase is present?

(HINT: See Lever Rule file)

- e. At 1300°C, what phases are present?
- f. At 1300°C, what percent of each phase is present?
- f. What phases are present at 1200°C?
- g. At 1200°C what is the percent of each phase present?
- h. What temperature does the last liquid disappear?
- i. What is the composition of the last liquid? \_\_\_\_\_\_\_\_



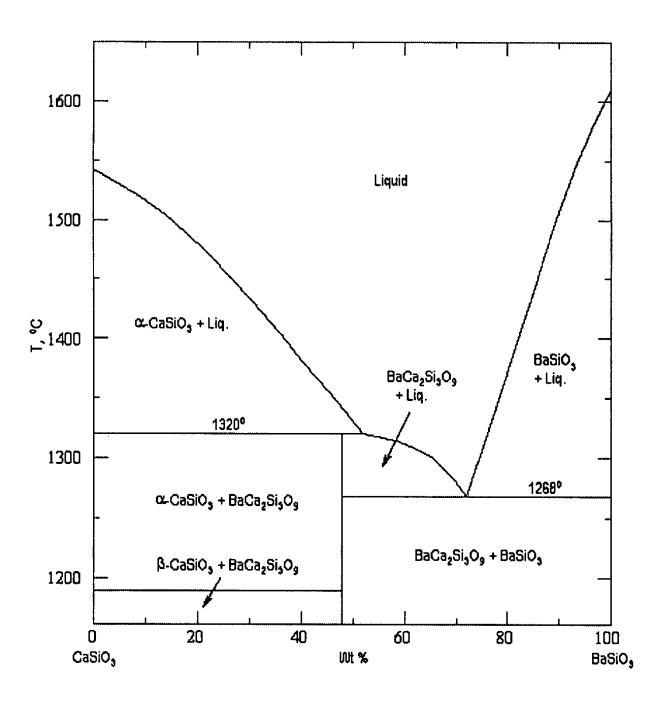


Figure 1